

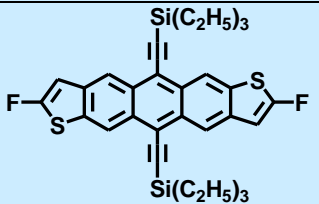
## Pentafluorothiophenol (abb ; PFTP) C<sub>6</sub>F<sub>5</sub>-SH

Purity	97%
CAS Number	771-62-0
Molecular Formula	C <sub>6</sub> HF <sub>5</sub> S
Molecular Weight	200.13

1. Self-oriented single-layer PFTP is coated on Cu, Ag, and Au electrodes to improve hole mobility and contact resistance of organic FETs. PFTP-Cu is better than bare Au electrodes.

diF-TES-ADT

Org.FETstructure(Red; PFTP)

diF-TES-ADT			
Au, Ag, Cu	Au, Ag, Cu		
SiO <sub>2</sub>			
p <sup>+</sup> -Si			
No.	Electrode	Hole mobility(cm <sup>2</sup> V <sup>-1</sup> S <sup>-1</sup> )  V <sub>GS</sub> -V <sub>T</sub>  =30V	Contact resistance (Ω · cm) V <sub>GS</sub> -V <sub>T</sub>  =30V
1	Bare-Ag	1.2*10 <sup>-4</sup>	8.0*10 <sup>7</sup>
2	Bare-Au	1.0*10 <sup>-2</sup>	3.0*10 <sup>7</sup>
3	PFTP-Cu	3.1*10 <sup>-2</sup>	2.0*10 <sup>6</sup>
4	PFTP-Ag	<b>1.2*10<sup>-1</sup></b>	3.2*10 <sup>5</sup>
5	PFTP-Au	<b>1.3*10<sup>-1</sup></b>	8.0*10 <sup>4</sup>

Application

*Applied Materials Interface, 2, 2015, 1400384*

2. Used for the preparation of gold particles with perfluorothiolate ligands. While the perfluoroalkylthiolate single-layer coated gold particles of the Brust reaction method are soluble in the fluorus solvent, the TFTP-coated gold particles exhibit the property of being soluble in the hydrocarbon solvent.

Au<sub>55</sub>(PPh<sub>3</sub>)<sub>12</sub>Cl<sub>6</sub> + excess-C<sub>6</sub>F<sub>5</sub>SH in CH<sub>2</sub>Cl<sub>2</sub>/RT\*22hr → evaporate → CH<sub>3</sub>OHwash/filtrate → Au<sub>75</sub>Gold particle (CH<sub>2</sub>Cl<sub>2</sub>, DMF, THF soluble)

*Langmuir, 2008, 24, p310-315*

### Properties:

Appearance	Liquid
Boiling point, °C	143-144
Melting point, °C	-24

Capacity:	150kg/month
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Packing:	-
UN, PG:	-